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LabChemistry_IS2_20160309_Camera1_Seg01.pdf

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Setting: IS2 instructs students about the experiment they are about to run.

Participants: IS2 (male, unseen), S1 (male student, unseen), S2 (male UGTA, unseen), S3 (female UGTA, unseen), S4 (female student, unseen), S5 (another female student, unseen)

(0:00)

```
XXX IS2: uh solvent
          whatever you use
XXX
XXX
          because
XXX
          in the first part
XXX
          you (need to) use uh hexene
          pure hexene
XXX
XXX
          and pure acyl acetate
XXX
          in a one to one ratio
XXX
          hexene to acyl acetate
XXX
          make sure
XXX
          that all these three solvent
XXX
          you need a thr- ah
XXX
          (ten mil)
XXX
          so for
XXX
          for example
          for this part you need five mil hexene
XXX
XXX
          and five mil acyl acetate
XXX
          because
XXX
          don't let this (height too high) -
XXX
          uh don't let this (height too high)
          and uh this is one
XXX
          uh:
XXX
          height requirement
XXX
XXX
          and another one is here
XXX
          so
XXX
          I think you
XXX
          maybe
XXX
          you need a ruler
XXX
          uh for the first time
          to do the (TLC)
XXX
XXX
          make sure
XXX
          the height is here is
XXX
          uh:
          one centimeter (.1)
XXX
XXX
          ok,
```

```
XXX
          so
XXX
          just make sure after you put this uh
XXX
          TLC
XXX
          in your chamber (.3)
XXX
          make sure uh
XXX
          your-
XXX
          your- your- your sample
XXX
          is higher
          than the solvent
XXX
XXX
         than the- than than
         than the other solvent
XXX
XXX
         ok,
         because
XXX
XXX
          if your solvent
         is underneath the solvent
XXX
XXX
         your solvent will dissolve
XXX
          in the- in the
XXX
          y- your h-
XXX
          your sample will dissolve in the solvent
XXX
          (it will) do not go up by the solvent
XXX
          ok,
XXX
          make sure the s-
XXX
          let's do a height requirement
XXX S1: yeah sure
         ((pause))
XXX
XXX IS2: so uh nothing is uh
XXX
        so for all-
          for all the uh glassware today
XXX
          don't wash with acetone
XXX
XXX
          or water
          even if it's not quite clean
XXX
XXX
          just use uh
XXX
         (methylene) chloride
          don't use acetone
XXX
          don't use uh
XXX
XXX
          water
XXX
          to wash all your glassware today
XXX S2:
         [uh
XXX IS2: [so yeah yeah
XXX S2: can I,
XXX IS2: yeah yeah
XXX S2:
          guys
XXX
          one thing I forgot to say is
```

```
the first part of the experiment
XXX
          you will be working in groups of three,
XXX
XXX
          right,
XXX
          and ((indistinguishable))
XXX
          just pick your-
XXX
          ok=
XXX S3:
          =just [pick people] who are around you
XXX S2:
              [yeah
                          1
XXX
          so that- that means uh one person
XXX
          does a certain type of solvent
XXX
          everyone understands that,
          make sure you write your partner's
XXX
          first and last name=
XXX
XXX S3:
          =yes=
XXX S2: =in your lab notebook
XXX
        please=
XXX S3: =at least two partners
XXX
          well
XXX
          yeah
          you only have [two partners
XXX
XXX S2:
                        [yeah
XXX
          you have two partners,
          write their names down in your lab notebook
XXX
XXX
          that will most likely be a quiz question=
          =yeah [that was my quiz question
XXX S3:
XXX S2:
               [I know
XXX
        it was my notebook quiz question as well
XXX
         ok,
XXX IS2: so
XXX S2: yeah
XXX IS2: so anoth- another
XXX
        last w-
XXX
         last uh:
          last thing I need to
XXX
          uh explain to you is
XXX
XXX
          uh
          about this plate
XXX
XXX
          so
          uh in total we have
XXX
XXX
          uh
        six sample
XXX
          to uh uh to run
XXX
XXX
         to run the (TLC) today
```

```
XXX
          so
XXX
          uh
XXX
          I just
XXX
          I want to remind you
XXX
          what's this uh
XXX
          three spot
XXX
          so
XXX
          because it kinda confusing
          so when I first have TA
XXX
XXX
         uh last semester
XXX
          so
         this is uh:
XXX
        standard
XXX
          so make sure (.) it is
XXX
         because we have two uh
XXX
XXX
          samples today
XXX
          we need uh
XXX
          for the part b we have a reaction
          so the reaction is uh
XXX
XXX
          ((pause))
XXX
          is (relate) to the uh
XXX
         fumerate (.2)
XXX
          so
          under uh
XXX
          bromine and cataly-
XXX
XXX
          catalyze reaction
XXX
          SO
XXX
          so f- for the uh
XXX
          first part
XXX
          make sure you use fumerate
          not (malate)
XXX
XXX
          ok,
(3:00)
          so this is the first standard point
XXX
          so this is a product
XXX
XXX
          this is your uh:
XXX
          starting material
XXX
          because we want to use TLC to monitor
          the reaction
XXX
         for this reaction
XXX
        so for the first point (.)
XXX
         use the fumerate
XXX
XXX
         not
```

```
XXX
          uh
XXX
          malate
XXX
          ok,
XXX
          and for the
XXX
        zero
        zero minutes
XXX
XXX
         now which means uh
        before reaction starts
XXX
         you need to uh
XXX
XXX
         (spot)
         uh for the zero uh
XXX
XXX
        for the zero minute
XXX
        which means uh you have uh
          flask
XXX
XXX
         you put your uh
XXX
        malate
         this is starting material
XXX
          inside it
XXX
          and also you need the solvent
XXX
XXX
          inside it
          and then mix them
XXX
XXX
          so
XXX
          before you add bromine
XXX
          uh
XXX
          using your micropipette
XXX
         get some uh
XXX
          s- s-
XXX
          get some uh- some uh-
          so- uh sample
XXX
XXX
         uh
          to (spot) here
XXX
XXX
         so this is the zero point
XXX
        if
XXX
        so for a- uh
XXX
         the three uh
XXX
        this
XXX
          uh:
XXX
          for this point is
XXX
          uh
XXX
         after you add the bromine
        inside your flask
XXX
          after- so you need to record time
XXX
XXX
          after two minutes
```

```
you need to get some sample from your flask
XXX
XXX
           to (spot) the for the two minutes
XXX
           so
XXX
           ten is for ten minutes
XXX
           twenty is for twenty minutes
XXX
           and
XXX
           (about this) uh:
XXX
           crystals
           so this crystals y-
XXX
XXX
           so
           you will get this (clus) of crystals
XXX
XXX
           follow- following the: (.) procedure
XXX
           after you got your crystals
XXX
           for this one
XXX
           so you can-
XXX
           you can uh use uh crystallization
XXX
           to get this product
XXX
           after you get this solid
XXX
           using uh
XXX
           using uh s-
XXX
           uh:
XXX
           methylene-
           methylene chloride
XXX
           to dissolve your product
XXX
XXX
           (add) a spot here
XXX
           then you go the whole spot
XXX
           to run the TLC
XXX
           to check whether your reaction
XXX
           is good or not
XXX
           whether you got your product
XXX
           or not
XXX
           because you have standard
XXX
           you can compare all the-
XXX
           all other
XXX
           spot
           with this standard
XXX
           because this standard is a-
XXX
XXX
           your product
XXX
           SO
XXX
           you can use TLC to m-
           to- to find whether
XXX
           your reaction is work or not
XXX
XXX
           ok, (.1)
```

```
XXX
          so uh
XXX
          (during uh) if you use bromine solution
XXX
          try to wear your
XXX
          uh
XXX
          heavy gloves
XXX
          because
XXX
          it is very toxic and uh
XXX
          uh
          erosive
XXX
XXX
          SO
          make sure you wear
XXX
          heavy gloves
XXX
XXX
          ok,
XXX S3:
          ((indistinguishable))
XXX
          latex-free gloves here
XXX
          but when you're using the bromine
XXX
          there's like heavy duty gloves
          near the actual solvent bench,
XXX
XXX
XXX
          but I wouldn't suggest putting your hands
XXX
          directly in them,
XXX
          wear gloves
          and then like
XXX
XXX
          use those as second gloves,
XXX
          yeah
XXX IS2: yeah
XXX
          and uh one more thing is uh
XXX
          so
          if you use micropipette
XXX
XXX
          so you y- y- y- you
          uh the thing is you need to transfer
XXX
XXX
          the solvent
XXX
          transfer your sample
XXX
          from y-
XXX
          from the micropipette
XXX
          to the TLC plate
XXX
          SO
XXX
          the thing is
          uh after you do the
XXX
          1- 1- the
XXX
          do the micropipette
XXX
          o- on your TLC
XXX
          don't
XXX
```

```
XXX
          uh just leave the top of your pipette open
XXX
          don't uh
XXX
          cover your
          pipette
XXX
XXX
          don't cover your micropipette
          don't cover the (.) top of your
XXX
XXX
          micropipette
XXX
          just do the
          uh just uh
XXX
XXX
          u:m
(6:00)
XXX
          spot- spot- spot where you have
          and where you here and here
XXX
XXX
          don't just cover the top of this
XXX
          ok,
XXX
          uh:
XXX
          so: yeah
          that's
XXX
         that's all that
XXX
XXX
          I have to say
XXX
          SO
XXX
          yeah,
XXX S1: ((indistinguishable))
XXX IS2: what,
XXX S1: ((indistinguishable))
XXX IS2: I think uh
XXX S1: ((indistinguishable))
XXX IS2: they say email it
        to all of them
XXX
XXX S1:
        ok
XXX IS2: so
XXX
        so uh because we have uh
XXX
         some uh
          research today
XXX
          about how to
XXX
XXX
          uh
          international TA interact with uh
XXX
XXX
         your students
XXX
          so
XXX
         um uh
         be- uh
XXX
         so just
XXX
XXX
         be uh
```

```
XXX
         relaxed
XXX
          don't
XXX
         because I am uh more nervous than you guys
XXX
         ok,
XXX all: ((laugh))
XXX IS2: just do experiment so
       experiment's your focus
XXX
        don't focus any other things
XXX
XXX
          ok,
XXX
        so make sure you can finish=
XXX S3: =yeah=
XXX IS2: =uh today's reaction
XXX S2: yeah
XXX just tune them out
XXX S3: =((indistinguishable))=
XXX S2: yeah
        also
we're here for you
XXX
XXX
        we believe in you
XXX
XXX ((various people laugh))
XXX IS2: yeah
XXX
       ok
XXX S4: so the um
XXX standard thing
XXX is the fumerate,
XXX S3: yeah
XXX IS2: yeah is the (.) product
        you fumerate
XXX
XXX
         not use malate
        because we have two standard-
XXX
         two things right here
XXX
XXX
         just
         don't (.) be confused
XXX
        about these two
XXX
XXX
        ok,
XXX S4: ok
XXX IS2: just follow your procedure
```

```
XXX
          you have
XXX
          so
          this is
XXX
XXX
          yeah
XXX
          yep
(7:12)
XXX
          ((pause))
(7:52)
XXX IS2: uh so guys
XXX
          uh
          crystallization report form is due today
XXX
XXX
          I will collect them
          uh
XXX
XXX S3:
          guys
          uh the crystallization report form
XXX
XXX
          on the benzoic acid product is due today
XXX
          forgot to mention that
(8:08)
XXX
          ((pause))
(8:34)
XXX IS2: ((to S2)) you'll keep this=
XXX S2: = [yeah]
XXX IS2: [on the pile
XXX S2: yeah
          I love this
XXX
          I love giving these out
XXX
(8:38)
XXX
          ((pause))
(9:33)
XXX S5:
          are you collecting,
XXX
          uh
XXX
         [our reports right now,
XXX IS2: [quiz,
         uh: no
XXX
         oh
XXX S5:
XXX IS2:
          yeah
(9:38)
XXX
          ((pause))
(10:57)
```